

<u>Thermal Energy</u>: is the total energy of motion in the particles of a substance.

Example: A teapot full of hot tea has more thermal energy than a cup full of hot tea because there are more particles in the pot.

Heat: is the transfer of thermal energy from a hotter object to a cooler one.

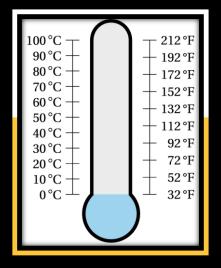






<u>Temperature</u>: is the average amount of energy of motion of each particle of a substance. (A measure of how hot or cold something is.)





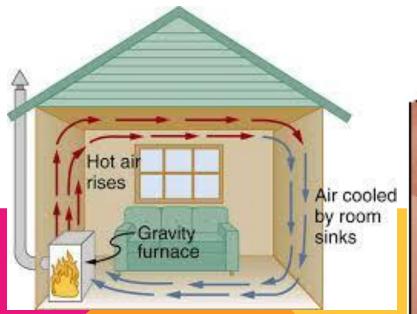
KEY IDEAS:

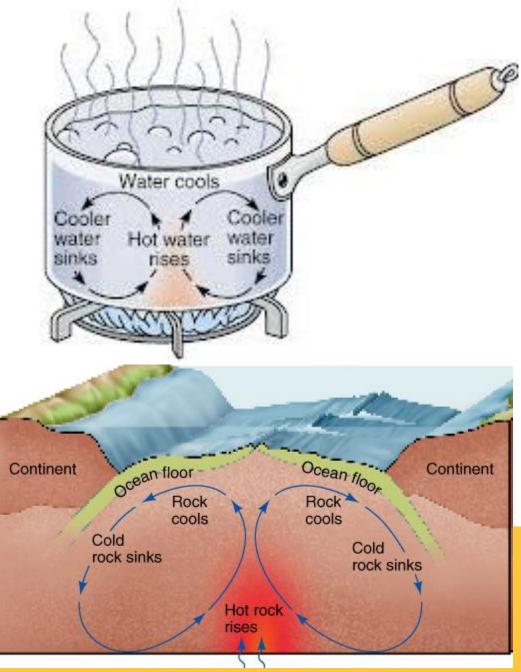
- Energy can move from one place, object, or system to another.
- Substances heat or cool as a result of energy transfer.
- Energy transfers from warmer matter to cooler matter until both reach the same temperature.
- Heat transfers occur in three ways: by radiation, conduction, and convection.



CONVECTION:

is the transfer of heat by the movement of a fluid.





CONDUCTION:

Is the direct transfer of heat from one substance to another substance that it is touching.



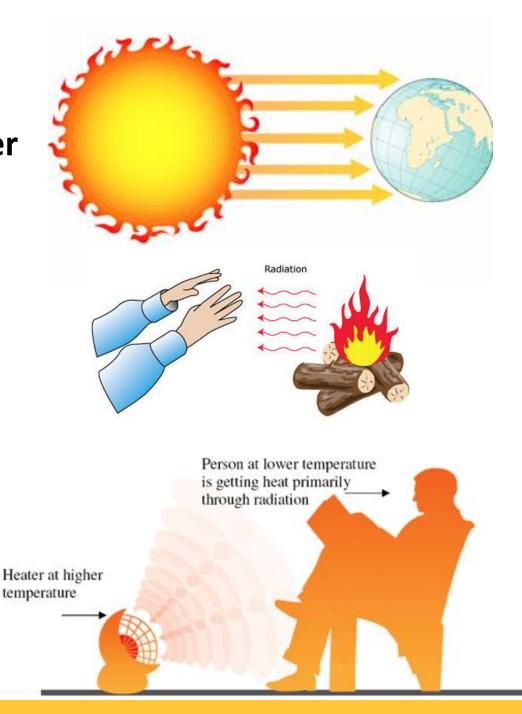


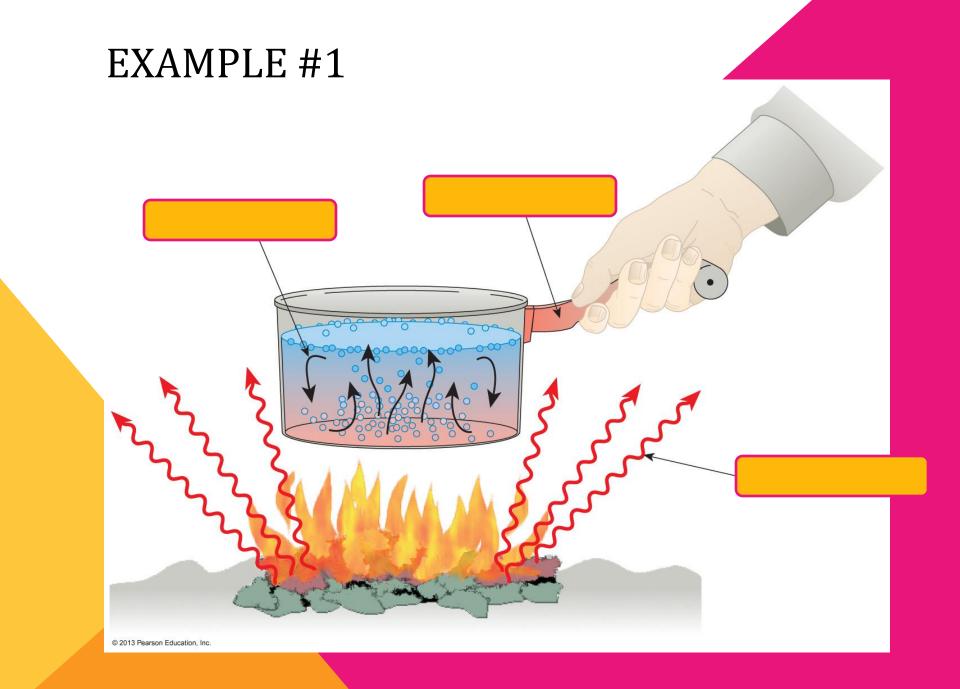


RADIATION:

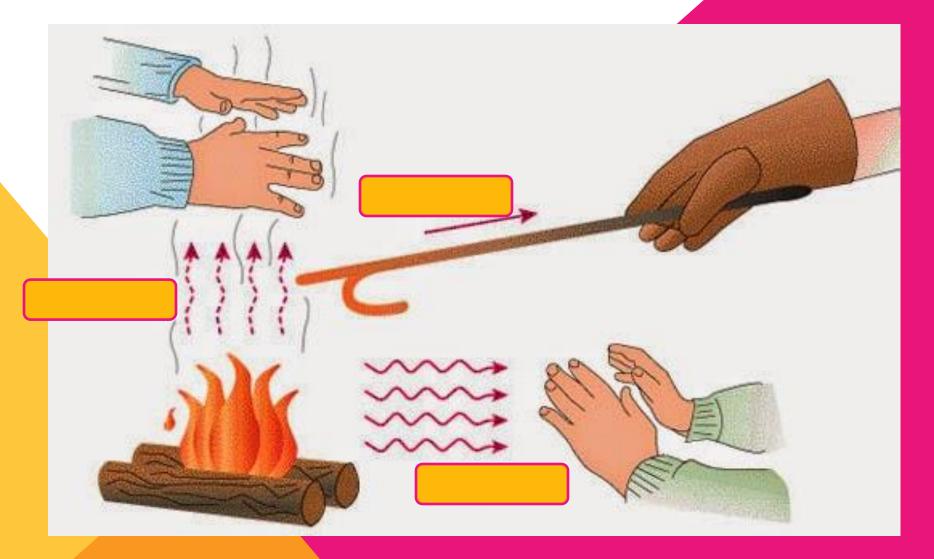
Is the direct transfer of energy by electromagnetic waves.

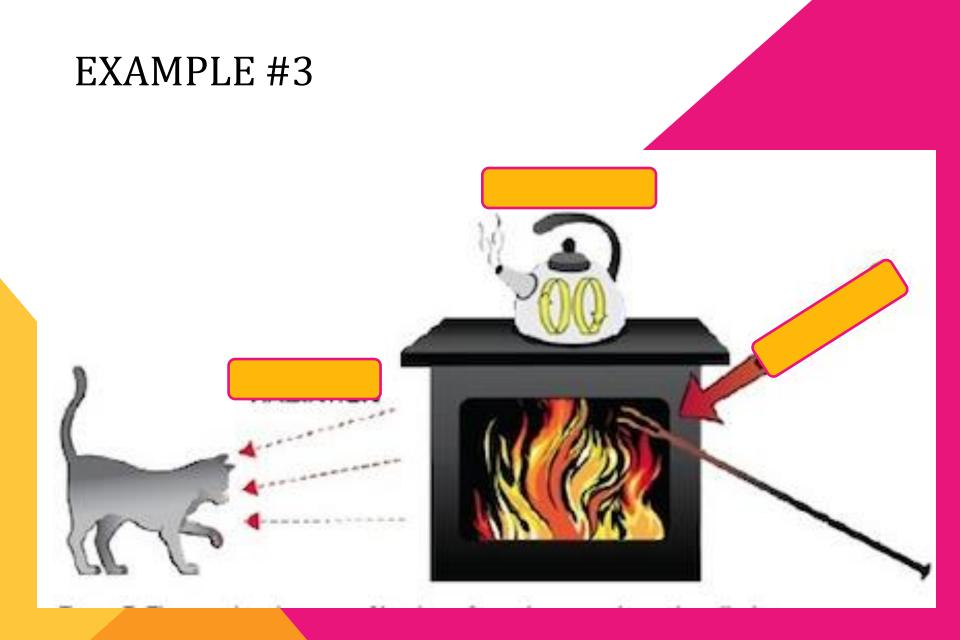






EXAMPLE #2





EXAMPLE #4

